

Available online at www.sciencedirect.com

Il Farmaco 58 (2003) 977-981

IL FARMACO

www.elsevier.com/locate/farmac

Synthesis and affinity for serotonin and dopamine transporters of some benzophenone oxime ether derivatives

Annalina Lapucci^{a,*}, Susanna Nencetti^a, Gian Carlo Demontis^b

^a Dipartimento di Scienze Farmaceutiche, Facoltà di Farmacia, Università di Pisa, Via Bonanno 6, 56100 Pisa, Italy
^b Facoltà di Farmacia, Dipartimento di Psichiatria, Neurobiologia, Farmacologia e Biotecnologie, Univer

Received 4 September 2002; accepted 12 December 2002

Abstract

In a previous study, we reported the synthesis of several 3-(methylenaminoxymethyl)-substituted piperidine derivatives and their ability to interfere with the transmission mediated by biogenic amines. The present study describes the preparation of some new oxime derivatives and their capacity to inhibit serotonin (SERT) and dopamine (DAT) transporters expressed at the level of the rabbit cortex and the striatal membranes, respectively. All the compounds showed K_i values in the micromolar range on both transporters.

 \odot 2003 Éditions scientifiques et médicales Elsevier SAS. All rights reserved.

Keywords: Dopamine; Serotonin; Oxime derivatives

1. Introduction

According to the monoamine theory, depression is a disease of the central nervous system associated with a dysfunction of the metabolism and/or of the synaptic transmission mediated by the biogenic amines dopamine (DA), serotonin (5-HT), noradrenaline (NA) [\[1,2\]](#page-4-0). The therapeutic treatment of this pathology is linked with the use of drugs that enhance the synaptic levels of these amines such as biogenic amines reuptake inhibitors. In this class of antidepressant agents, a particular importance is attached to selective serotonin reuptake inhibitors (SSRIs), such as fluoxetine, sertraline, paroxetine, citalopran and fluvoxamine, which have been used with success in depressed patients in view of their favourable side-effect profile with respect to classic tricyclic antidepressants (TCAs) [\[1,3\].](#page-4-0)

In a previous work [\[4\],](#page-4-0) we evaluated the inhibitory activity of some 3-(methylenaminoxy)methyl piperidines of type 1 ([Fig. 1](#page-1-0)) on biogenic amine reuptake (NA, DA, 5-HT) in rabbit brain synaptosomal fractions. This study showed that all compounds possess an appreciable activity, and the one in which R and R_1 are two phenyls (1a) is the most active in this series of compounds, with K_i values in the micromolar range. On the basis of these data, with the aim to gain further knowledge about the influence of certain molecular portions of 1a on its biological properties, we synthesised compounds 2a, 2b, 3a and 3b, in which the benzophenone oximic portion of 1a is still present, while the aminic side-chain, ethylpiperidinic (2a and 2b) or ethyldimethylaminic (3a and 3b), is different.

2. Chemistry

* Corresponding author. E-mail address: alapucci@farm.unipi.it (A. Lapucci).

The oxime ethers 2a, 2b, 3a and 3b were synthesised as indicated in [Scheme 1.](#page-1-0) Reaction of benzophenone or 4-

Fig. 1.

fluorobenzophenone with hydroxylamine hydrochloride afforded the oximes 4a and 4b [\[5,6\]](#page-4-0). The reaction of 4a and 4b with the appropriate alkyl chloride in DMSO and sodium hydroxide, in the presence of 18-Crown-6, yielded the crude oxime ethers 2a, 2b, 3a and 3b, which

were purified by column chromatography and then transformed into the corresponding hydrochloride salts.

The structure of compounds 2 and 3 was confirmed by their spectral data. The E configuration around the double bond of the methylenaminoxy group of com-

Scheme 1.

pounds 2b and 3b was assumed on the basis of the configuration of the starting oxime 4b [\[6\]](#page-4-0) which have been proved to be configurationally stable in the reaction conditions employed.

3. Pharmacology

Compounds 2a, 2b, 3a and 3b were tested for their ability to interfere with the systems of transmission of biogenic amines, by directly evaluating the degree to which they bind to the serotonin (SERT) and dopamine (DAT) transporters expressed at the level of the rabbit cerebral cortex and the striatal membranes, respectively. In order to allow a more homogeneous comparison of all biological data, the same type of test was also used for compound 1a, which was previously tested [\[4\]](#page-4-0) by means of the direct uptake method. $[^{3}H]$ -paroxetine was used as the specific tritiated ligand for SERT, while $[3H]$ WIN 35,428 was used to label DAT. The binding affinity indices of oxime ethers 2a, 2b, 3a and 3b are reported in Table 1, together with those obtained in the same tests with the previously studied compound 1a and fluoxetine.

4. Results and discussion

As regard affinity for SERT, the data in Table 1 show that all the synthesised compounds 2a, 2b, 3a and 3b exhibit very low affinity values with K_i in the micromolar range, in the same way as the previously studied compound 1a whereas the affinity value of the reference compound, fluoxetine is in the nanomolar range.

Also, the affinity for DAT of the compounds 2a and **3a**, in which $R = H$, is low with K_i values in the micromolar range, so the DAT/SERT affinity ratio is very low, whereas for the fluoxetine is 689.

Furthermore, it may be seen from Table 1 that the compounds in which one of the aromatic rings is substituted in the para position with a fluorine atom (2b and 3b) exhibit a SERT affinity index slightly lower than that of the corresponding compounds unsubstituted on the aromatic ring, without any appreciable change in their affinity for DAT, with a ratio of 4.1 for compound 3b.

In conclusion, it may be said that the changes made in the structure of compound 1a lead to a slight increase in the affinity for SERT, in particular in compounds in which $R = F(2b \text{ and } 3b)$. Overall results indicate that compounds 2a, 2b, 3a and 3b exhibit a certain selectivity for DAT compared with SERT, but this tendency is attenuated by the insertion of a fluorine atom on one of the aromatic rings.

5. Experimental

5.1. Chemistry

Melting points were determined on a Kofler hot stage apparatus and are uncorrected. IR spectra for comparison between compounds were recorded with a Perkin-Elmer model 1310, as nujol mulls in the case of solid substances, or as liquid film in the case of liquids. ${}^{1}H$ NMR spectra were routinely recorded with a Varian CFT 20 (80 MHz) instrument in ca. 5% solution of $CDCl₃$ (for neutral compounds or the free bases) or $D₂O$ (for the salts), using Me₄Si or Me₃Si(CH₂)₃SO₃Na as the internal standard, respectively. The oximes $4a$ and (E) -4b were prepared by the method described in literature. [\[5,6\].](#page-4-0) Analytical TLC were carried out on 0.25 mm layer silica gel plates containing a fluorescent indicator; spots were detected under UV light (254 nm). Column chromatography was performed using $70-230$ mesh silica gel. $MgSO₄$ was always used as the drying agent. Evaporation was carried out in vacuo (rotating evaporator). Elemental analyses were performed by our analytical laboratory and agreed with theoretical values to within $+0.4%$.

5.1.1. General procedure for the synthesis of oximethers $2a$, $2b$, $3a$ and $3b$

N-(2-Chloroethyl) piperidine or 2-(dimethylamino) ethyl chloride hydrochloride (23 mmol), NaOH (46 mmol) and 18-Crown-6 (23 mmol) were added to a stirred solution of the appropriate oxime 4a and 4b [\[5,6\]](#page-4-0) (23 mmol) in anhydrous DMSO (20 ml). After comple-

Table 1 Affinity for SERT and DAT of compounds 1a, 2a, 2b, 3a and 3b

Comp.	R	R_1	5-HT-T K_i (μ M)	DAT K_i (μ M)	Ratio
2a	Н	$bdc-C5H10N$	$11.4 + 2.7$	$3.6 + 1.2$	0.3
2 _b	F	$bdc-C5H10N$	$4.2 + 1.3$	3.9 ± 1.1	0.9
3a	Н	N(Me)	$9.0 + 3.1$	7.2 ± 2.6	0.8
3 _b	F	N(Me)	$1.2 + 0.3$	$4.9 + 1.4$	4.1
1a			$25.0 + 6.1$	n.t.	
Fluoxetine			$5.8 \times 10^{-3} \pm 2.9 \times 10^{-3}$	$4.0 + 1.7$	689

tion of the addition the reaction mixture was stirred at 40° C for 48 h, diluted with water and extracted with CHCl₃. The evaporation of the washed $(H₂O)$ and dried CHCl₃ extracts give an oily residue that was purified by column chromatography $(CHCl₃/5%$ MeOH) to yield pure 2a, 2b, 3a and 3b.

2a: (39%) ¹H NMR δ 1.25–2.40 (m, 10H), 2.58–2.73 $(m, 2H), 4.22-4.37$ $(m, 2H), 7.06-7.89$ $(m, 10H).$ 2b: (96%) ¹H NMR δ 1.43–2.34 (m, 10H), 2.52–2.65 (m, 2H), 4.18-4.40 (m, 2H), 7.01-7.43 (m, 9H). 3a: (59%) ¹H NMR δ 2.24 (m, 6H), 2.57-2.72 (m, 2H), 4.20-4.36 (m, 2H), 7.29–7.39 (m, 10H). **3b:** (37%) ¹H NMR δ 2.25 $(m, 6H), 2.48-2.72$ $(m, 2H), 4.19-4.36$ $(m, 2H), 6.96-$ 7.37 (m, 9H).

Compounds 2a, 2b, 3a and 3b were converted to the corresponding hydrochloride salts by dissolving the bases in $Et₂O$ and subsequent treatment with an excess of $Et₂O·HCl$. The solid precipitate was filtered and crystallised from MeOH/Et₂O to yield pure $2a$ HCl, $2b$. HCl, 3a HCl, 3b HCl. Melting points are reported in Table 2.

5.2. Binding studies

5.2.1. Affinity for 5-HT transporter (SERT)

The affinity of compounds (2a, 2b, 3a, and 3b) for the serotonin transporter was assayed in rabbit cortex membranes, labelled with [³H]-paroxetine, whose specific activity ranged from 15 to 20 Ci/mmol.

Tissue from the anterior pole of the rabbit frontal cortex was collected and stored at -35 °C until used. Membranes were prepared from frozen tissue using an ultraturrax homogeniser, in a saline solution whose composition was (mM) : NaCl (120) ; KCl (5) , Tris-HCl (50), pH 7.4 at 4° C, using a dilution factor of 1:30 (w:v). The homogenate was centrifuged for 10 min at a relative centrifugal force of $46,000 \times g$, keeping the temperature at 4° C. The supernatant was discarded and the pellet was resuspended by diluting 1:30 (w:v) in the homogenisation buffer. The pellet was recentrifuged at $46,000 \times g$ for 10 min and the pellet resuspended 1:30 in buffer without sucrose, splitted in 1 ml aliquots and stored frozen at -35 °C. On the day of the experiment, one aliquot was quickly thawed at $37 \degree C$, centrifuged for 10 min at $46,000 \times g$ at 4° C and the pellet resuspended in homogenisation buffer at a dilution of 1:30 (w:v).

Competition assays were performed in duplicate glass tubes in a total volume of $2000 \mu l$, with a concentration of $[^{3}H]$ -paroxetine of 100–200 pM, using an incubation time of 2 h at room temperature. In the experimental conditions used, specific binding (see below) reached equilibrium after 90 min. Separation of ligand bound to the transporter from free ligand was performed by filtration using a 30 well manyfold (Brandel), with glass fiber filters (GF/C) pre-soaked for 2 h at room temperature with 0.6% (w:v) polyethylenimine. Filters were quickly washed four times with cold $(4^{\circ}C)$ Tris-HCl 50 mM. Filters were removed, put into scintillation vials and let to stay overnight in 3 ml of Cytoscint ES (ICN). The following day, samples were read by scintillation spectroscopy in a β -counter (Packard). Unspecific binding, defined in the presence of 2 mM serotonin, was subtracted from total binding (in the absence of competitors) to obtain specific binding. In a typical assay, total binding was 900 d.p.m., unspecific was 200 d.p.m., and specific binding was 700 d.p.m.

5.2.2. Affinity for DA transporter (DAT)

The affinity of compounds (2a, 2b, 3a, and 3b) for the DAT was assayed in rabbit striatal membranes, labelled with the cocaine analogue $[3H]$ -WIN 35,428, whose specific activity was 84.5 Ci/mmol.

Tissue was collected and stored at -35 °C until used. Membranes were prepared from frozen tissue using an ultraturrax homogeniser, in a saline solution whose composition was (mM): NaCl (48); sucrose (320), pH 7.7 at 4 °C, by mixing $Na₂HPO₄/NaH₂PO₄$ (25). The homogenate was centrifuged for 10 min at 48,000 relative centrifugal force of $46,000 \times g$, keeping the temperature at 4° C. The supernatant was discarded and the pellet resuspended by diluting 1:100 (w:v) in the homogenisation buffer. The pellet was recentrifuged at $46,000 \times g$ for 10 min and the pellet was resuspended 1:100 in buffer without sucrose, splitted in 3.5 ml aliquots and stored frozen at -35 °C. On the day of the experiment, one 3.5 ml aliquot was quickly thawed at 37 °C, centrifuged for 10 min at $46,000 \times g$ at 4 °C and the pellet was resuspended in buffer without sucrose at a dilution of 1:200 (w:v).

Competition assays were done in duplicate glass tubes in a total volume of 500 μ l, with a concentration of $[^3H]$ -WIN 35,428 of $0.5-2$ nM, using an incubation time of 2

h at 4° C. In the experimental conditions used, specific binding (see below) reached equilibrium after 60 min. Separation of ligand bound to the transporter from free ligand was by filtration using a 30 well manyfold (Brandel), with glass fiber filters (GF/C) pre-soaked for 2 h at room temperature with 0.6% (w:v) polyethylenimine. Filters were quickly washed three times with cold $(4 °C)$ Tris-HCl 50 mM. Filters were removed, put in scintillation vials and let to stay overnight in 3 ml of Cytoscint ES (ICN). The following day, samples were read by scintillation spectroscopy in a β -counter (Packard). Unspecific binding, defined in the presence of 10 mM dopamine, was subtracted from total binding (in the absence of competitors) to obtain specific binding. In a typical assay, total binding was 600 d.p.m., unspecific was 150 d.p.m., and specific binding was 450 d.p.m.

The affinity of $[^{3}H]$ -WIN 35,428, for the dopamine transporter was assessed in separate saturation experiments, and gave an average dissociation constant (K_d) of 6.1 nM.

References

- [1] L.D. Green, K. Dawkins, in: M.E. Wolff (Ed.), Burger's Medicinal Chemistry and Drug Discovery, vol. 5, 5th ed., Wiley, New York, 1997, p. 121.
- [2] C.L.E. Broekkamp, D. Leysen, B.W.M.M. Peeters, R.M. Pinder, J. Med. Chem. 38 (1995) 4615-4633.
- [3] B. Oliver, W. Soudijn, I. Van Wijngaarden, Prog. Drug Res. 54 (2000) 59.
- [4] A. Balsamo, A. Lapucci, A. Lucacchini, M. Macchia, C. Martini, C. Nardini, S. Nencetti, Eur. J. Med. Chem. 29, 967-973.
- [5] A. Vogel, Vogel's Textbook of Practical Organic Chemistry, Longman, New York, 1978, p. 811.
- [6] G. Massolini, M.L. Carmellino, A. Baruffini, Farmaco 2 (1987) $117 - 124.$